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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.040 wR factor = 0.115 Data-to-parameter ratio = 14.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(E)-2-Benzoyl-4-[(4-methylphenyl)diazenyl]phenol

The title compound, $C_{20}H_{16}N_2O_2$, displays a *trans* configuration with respect to the N=N double bond. The aromatic rings bridged by the azo group are nearly coplanar, forming a dihedral angle of 6.83 (8)°. A strong intramolecular O– H···O hydrogen bond is observed. In the three-dimensional network, the molecules are linked by weak van der Waals interactions.

Comment

The azo compound class accounts for 60–70% of all dyes. These compounds are widely used in the textile, printing, paper manufacturing, pharmaceutical and food industries. All of them contain at least one azo group (-N = N-) which links two sp^2 -hybridized C atoms. During the course of our studies aimed at the synthesis and characterization of new azo compounds, the title compound, (I), was isolated. We report here the crystal structure of (I).



In the azo group of (I), the N1–C5 and N2–C8 bond lengths indicate significant single-bond character, whereas the N1–N2 bond length is indicative of significant double-bond character (Table 1). The benzene rings C2–C7 and C8–C13 adopt a *trans* configuration about the azo group, and are essentially coplanar, forming a dihedral angle of 6.83 (8)°. The dihedral angle between the C8–C13 and C15–C20 aromatic rings is 58.97 (8)°. A strong intramolecular O–H···O hydrogen bond is observed (Table 2). In the crystal structure, the molecules are linked by weak van der Waals interactions.

Experimental

A mixture of 4-methylaniline (0.54 g, 5 mmol), water (20 ml) and concentrated hydrochloric acid (1.25 ml, 15 mmol) was stirred until a clear solution was obtained. This solution was cooled to 273–278 K and a solution of sodium nitrite (0.41 g, 7 mmol) in water was added dropwise while the temperature was maintained below 278 K. The resulting mixture was stirred for 30 min in an ice bath. 2-Hydroxy-benzophenone (1 g, 5 mmol) solution (pH 9) was gradually added to

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a cooled solution of 4-methylbenzenediazonium chloride, prepared as described above, and the resulting mixture was stirred at 273–278 K for 60 min in an ice bath. The product was recrystallized from ethyl alcohol to obtain the solid title compound. Crystals suitable for X-ray analysis were obtained after 1 d by slow evaporation of an ethanol solution (yield 25%; m.p. 406–408 K).

 $D_{\rm x} = 1.324 {\rm Mg m}^{-3}$

Cell parameters from 17500

Mo $K\alpha$ radiation

reflections

 $\theta = 2.2-27.9^{\circ}$ $\mu = 0.09 \text{ mm}^{-1}$

T = 293 (2) K

Prism, brown

 $0.66 \times 0.46 \times 0.20 \text{ mm}$

Crystal data

 $\begin{array}{l} C_{20}H_{16}N_2O_2\\ M_r = 316.35\\ \text{Monoclinic, } P2_1/c\\ a = 15.6711 \ (13) \ \text{\AA}\\ b = 5.8068 \ (2) \ \text{\AA}\\ c = 18.7764 \ (13) \ \text{\AA}\\ \beta = 111.706 \ (6)^\circ\\ V = 1587.48 \ (18) \ \text{\AA}^3\\ Z = 4 \end{array}$

Data collection

2171 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.038$
$\theta_{\rm max} = 26.0^{\circ}$
$h = -19 \rightarrow 19$
$k = -7 \rightarrow 7$
$l = -22 \rightarrow 22$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0643P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.040$	+ 0.0625P]
$wR(F^2) = 0.115$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} < 0.001$
3115 reflections	$\Delta \rho_{\rm max} = 0.17 \ {\rm e} \ {\rm \AA}^{-3}$
217 parameters	$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected bond lengths (Å).

N1-N2	1.2522 (18)	N2-C8	1.423 (2)
N1-C5	1.424 (2)		

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
O1−H1···O2	0.82	1.84	2.5608 (17)	146

All H atoms were placed in calculated positions and refined using a riding model, with C-H = 0.93–0.96 Å, O-H = 0.82 Å and $U_{\rm iso}({\rm H})$ = 1.2–1.5 $U_{\rm eq}$ (parent atom).



Figure 1

A view of the molecular structure of (I). The intramolecular $O-H\cdots O$ hydrogen bond is indicated by a dashed line.



Figure 2

Packing diagram of (I), viewed along the *b* axis. Intramolecular $O-H \cdots O$ hydrogen bonds are shown as dashed lines.

Data collection: X-AREA (Stoe & Cie, 2002); cell refinement: X-AREA; data reduction: X-RED32 (Stoe & Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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