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## Structure Reports

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## (E)-2-Benzoyl-4-[(4-methylphenyl)diazenyl]phenol

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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.040$
$w R$ factor $=0.115$
Data-to-parameter ratio $=14.4$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]The title compound, $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$, displays a trans configuration with respect to the $\mathrm{N}=\mathrm{N}$ double bond. The aromatic rings bridged by the azo group are nearly coplanar, forming a dihedral angle of $6.83(8)^{\circ}$. A strong intramolecular $\mathrm{O}-$ $\mathrm{H} \cdots \mathrm{O}$ hydrogen bond is observed. In the three-dimensional network, the molecules are linked by weak van der Waals interactions.

## Comment

The azo compound class accounts for $60-70 \%$ of all dyes. These compounds are widely used in the textile, printing, paper manufacturing, pharmaceutical and food industries. All of them contain at least one azo group ( $-\mathrm{N}=\mathrm{N}-$ ) which links two $s p^{2}$-hybridized C atoms. During the course of our studies aimed at the synthesis and characterization of new azo compounds, the title compound, (I), was isolated. We report here the crystal structure of (I).

(I)

In the azo group of (I), the $\mathrm{N} 1-\mathrm{C} 5$ and $\mathrm{N} 2-\mathrm{C} 8$ bond lengths indicate significant single-bond character, whereas the $\mathrm{N} 1-\mathrm{N} 2$ bond length is indicative of significant double-bond character (Table 1). The benzene rings $\mathrm{C} 2-\mathrm{C} 7$ and $\mathrm{C} 8-\mathrm{C} 13$ adopt a trans configuration about the azo group, and are essentially coplanar, forming a dihedral angle of $6.83(8)^{\circ}$. The dihedral angle between the $\mathrm{C} 8-\mathrm{C} 13$ and $\mathrm{C} 15-\mathrm{C} 20$ aromatic rings is $58.97(8)^{\circ}$. A strong intramolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bond is observed (Table 2). In the crystal structure, the molecules are linked by weak van der Waals interactions.

## Experimental

A mixture of 4-methylaniline ( $0.54 \mathrm{~g}, 5 \mathrm{mmol}$ ), water ( 20 ml ) and concentrated hydrochloric acid ( $1.25 \mathrm{ml}, 15 \mathrm{mmol}$ ) was stirred until a clear solution was obtained. This solution was cooled to 273-278 K and a solution of sodium nitrite $(0.41 \mathrm{~g}, 7 \mathrm{mmol})$ in water was added dropwise while the temperature was maintained below 278 K . The resulting mixture was stirred for 30 min in an ice bath. 2-Hydroxybenzophenone ( $1 \mathrm{~g}, 5 \mathrm{mmol}$ ) solution ( pH 9 ) was gradually added to

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a cooled solution of 4-methylbenzenediazonium chloride, prepared as described above, and the resulting mixture was stirred at 273-278 K for 60 min in an ice bath. The product was recrystallized from ethyl alcohol to obtain the solid title compound. Crystals suitable for X-ray analysis were obtained after 1 d by slow evaporation of an ethanol solution (yield $25 \%$; m.p. 406-408 K).

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{20} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2} \\
& M_{r}=316.35 \\
& \text { Monoclinic, } P 2_{1} / c \\
& a=15.6711(13) \AA \\
& b=5.8068(2) \AA \\
& c=18.7764(13) \AA \\
& \beta=111.706(6)^{\circ} \\
& V=1587.48(18) \AA^{\circ} \\
& Z=4
\end{aligned}
$$

$$
\begin{aligned}
& D_{x}=1.324 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 17500 \\
& \quad \text { reflections } \\
& \theta=2.2-27.9^{\circ} \\
& \mu=0.09 \mathrm{~mm}^{-1} \\
& T=293(2) \mathrm{K} \\
& \text { Prism, brown } \\
& 0.66 \times 0.46 \times 0.20 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Stoe IPDS-2 diffractometer
2171 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.038$
$\theta_{\text {max }}=26.0^{\circ}$
$h=-19 \rightarrow 19$
$k=-7 \rightarrow 7$
$l=-22 \rightarrow 22$
15552 measured reflections

## Refinement

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0643 P)^{2}\right. \\
& +0.0625 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.17 \mathrm{e} \mathrm{~A}^{-3} \\
& \Delta \rho_{\text {min }}=-0.13 \mathrm{e}^{-3}
\end{aligned}
$$

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.040$
$w R\left(F^{2}\right)=0.115$
$S=1.02$
3115 reflections
217 parameters
H-atom parameters constrained


Figure 1
A view of the molecular structure of (I). The intramolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bond is indicated by a dashed line.


Figure 2
Packing diagram of (I), viewed along the $b$ axis. Intramolecular $\mathrm{O}-$ $\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds are shown as dashed lines.

Data collection: $X$-AREA (Stoe \& Cie, 2002); cell refinement: $X-A R E A$; data reduction: $X$-RED32 (Stoe \& Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

## References

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